



UNIVERSITY EXAMINATIONS

EXAMINATION FOR THE AWARD OF DEGREE OF BACHELOR OF SCIENCE (CHEMISTRY, INDUSTRIAL CHEMISTRY, BIO, MATHEMATICS) BACHELOR OF EDUCATION (SCIENCE)

CHEM 332: ORGANIC CHEMISTRY III

STREAMS:

TIME: 2 HOURS

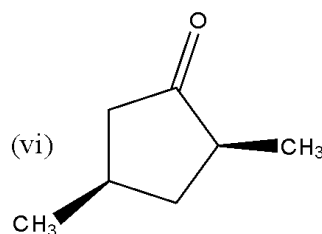
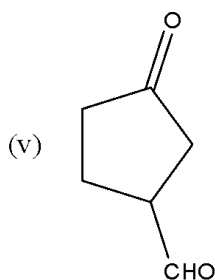
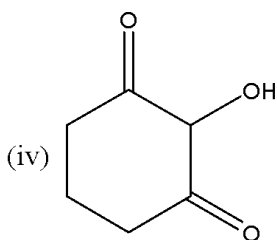
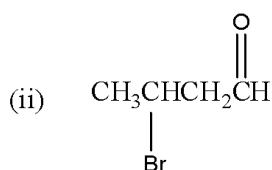
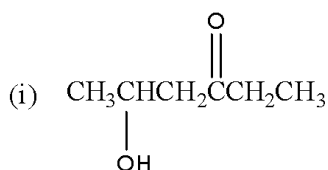
DAY/DATE: WEDNESDAY 18/12/2024

11.30 A.M – 1.30 P.M

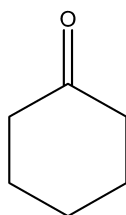
INSTRUCTIONS:

Answer question **One** (Compulsory) and any other **Two** questions**QUESTION ONE [30 MARKS]**

(a) Write the IUPAC name of each of the following organic compounds (3 marks)



(b) Write structures for the major organic product(s) formed when cyclohexanone is reacted with the following reagents. (6 marks)



Cyclohexanone

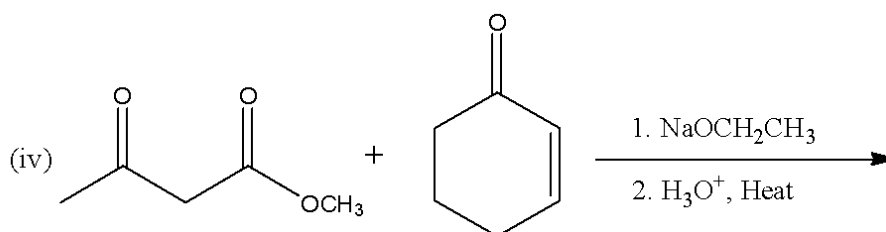
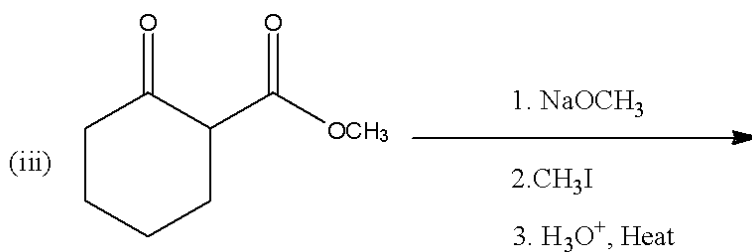
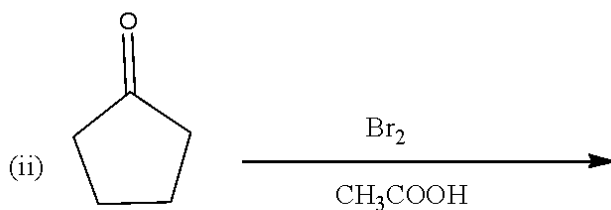
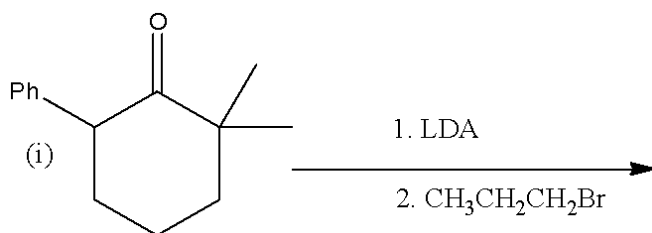


(iv) $\text{PhNHNH}_2, \text{H}_3\text{O}^+$ (v) PhMgBr then H_3O^+ (vi) NaCN, HCN then H_3O^+

(c) Discuss the physical properties of aldehydes and ketones.

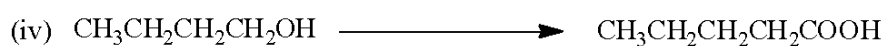
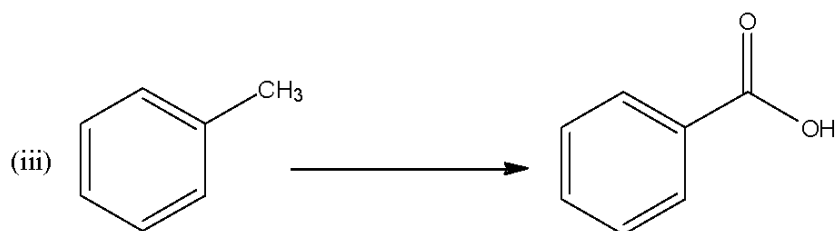
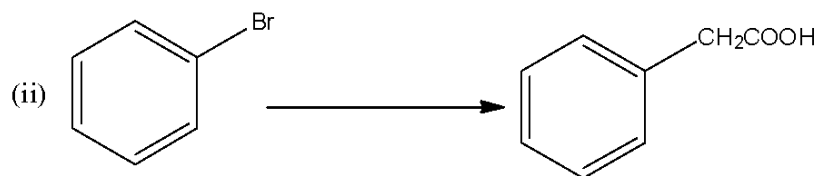
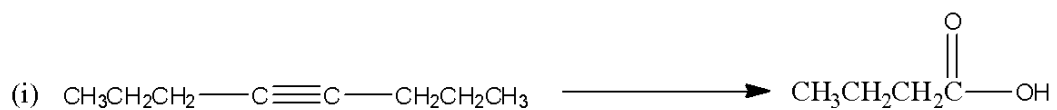
(3 marks)

(d) Write the major organic product(s) of the following reactions.

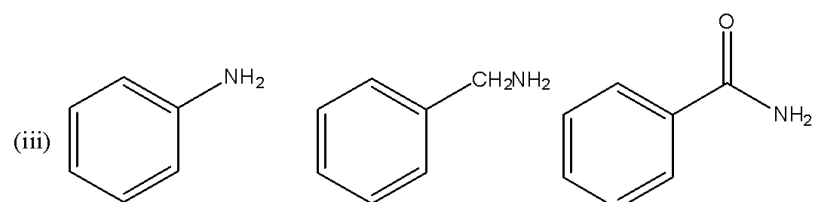
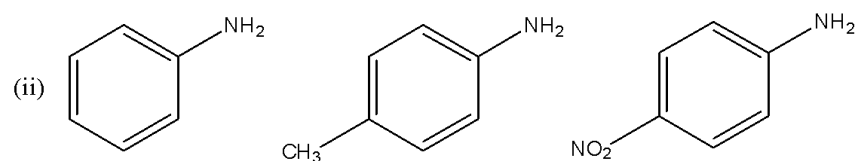
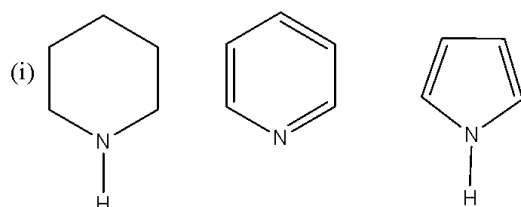
(6 marks)

(e) Provide the reagents that are required to effect each of the following transformations.

(6 marks)

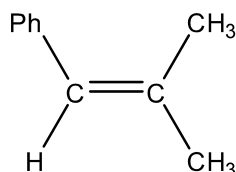


(f) Rank the amines in each set in order of increasing basicity. Justify your order in each case. **(6 marks)**

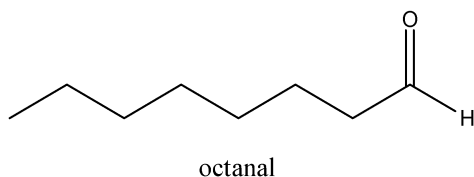


QUESTION TWO [20 MARKS]

(a) Show how the Wittig reaction can be used to synthesize the following alkene, starting with an alkyl halide and a ketone or an aldehyde. **(2 marks)**

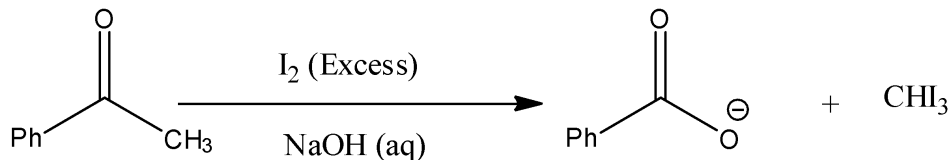


(b) Design a reasonable synthesis of octanal from each of the following compounds. **(6 marks)**



- (i) octan-1-ol (ii) non-1-ene (iii) oct-1-yne
 (iv) 1-bromoheptane (v) octanoic acid (vi) ethyl octanoate

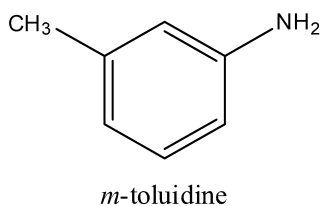
(c) Write the stepwise mechanism of the following reaction **(3 marks)**

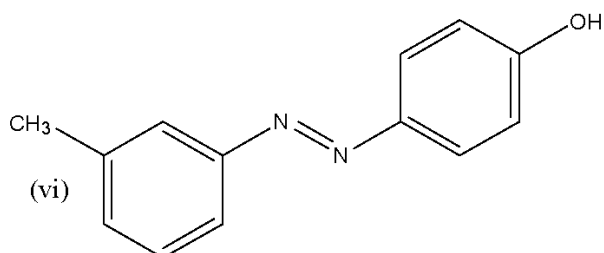
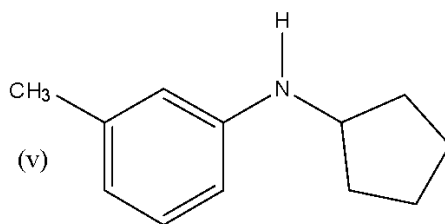
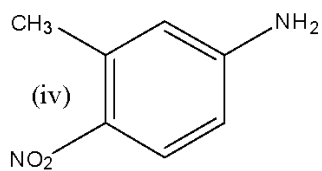
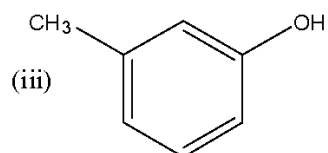
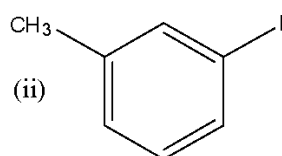
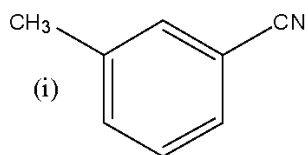


(d) Draw the structures of the following compounds **(3 marks)**

- (i) *tert*-butylamine (ii) N-ethyl-N-methylhexan-3-amine (iii) N-methylpyrrolidine
 (iv) N,N-diethylaniline (v) N-methyl-N-propylcyclopentanamine (vi)
 N,4-diethylhexan-3-amine

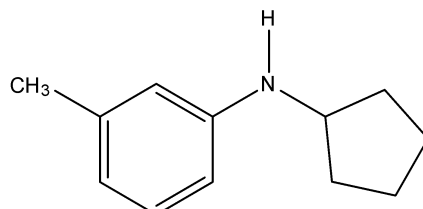
(e) Design a reasonable synthesis for conversion of *m*-toluidine to the following compounds. **(6 marks)**





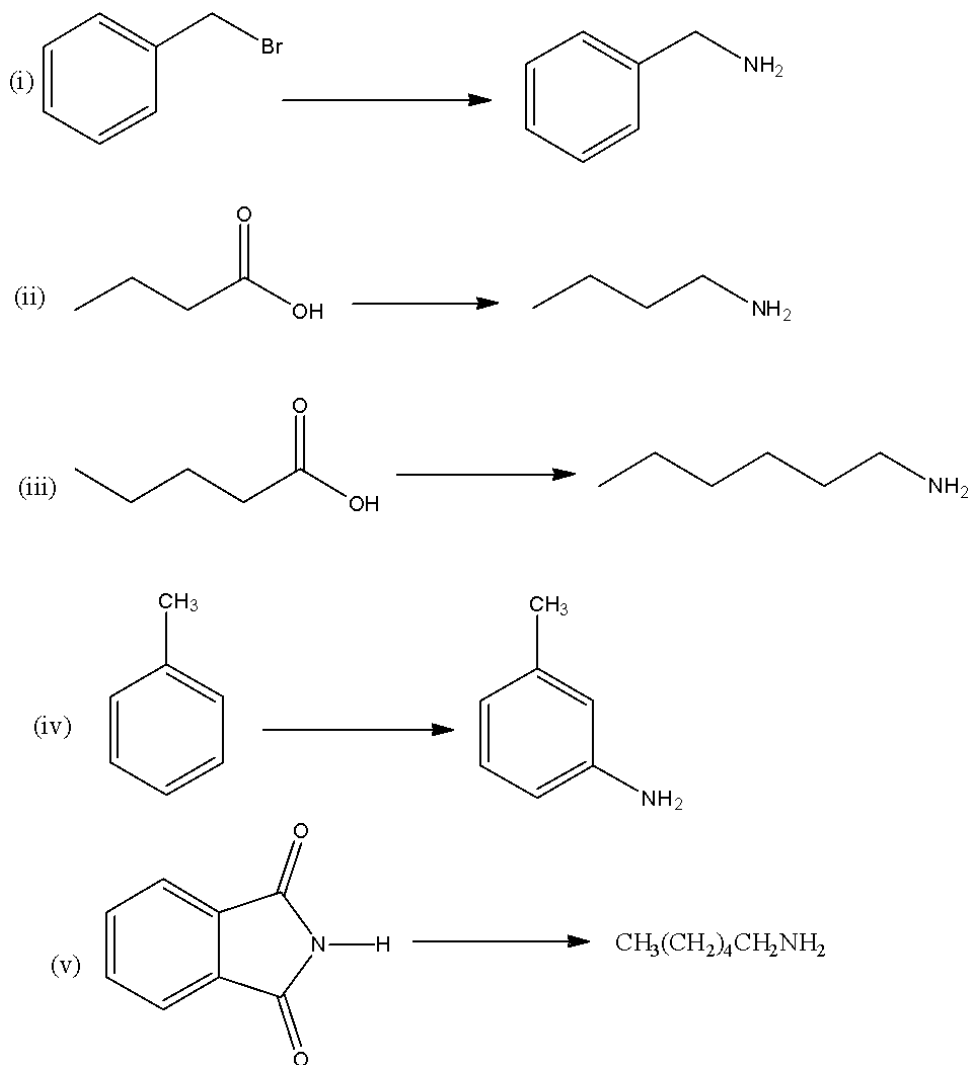
QUESTION THREE [20 MARKS]

(a) Design a reasonable synthesis of the following amine using reductive amination. **(2 marks)**

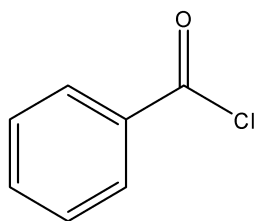


(b) Discuss the physical properties of amides. **(2 marks)**

(c) Provide the reagents that are required to effect each of the following transformations. **(10 marks)**



(d) Write the structure of the major organic product(s) of the reaction of n benzoyl chloride with each of the following reagents. **(6 marks)**



benzoyl chloride

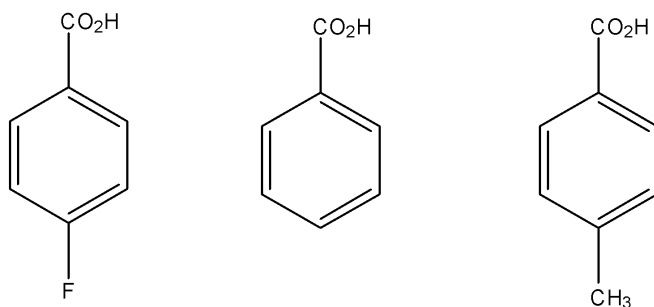
- (i) ethanol (ii) sodium acetate (iii) anisole and aluminum chloride
 (iv) excess CH_3MgBr , then dilute acid (v) $\text{LiAlH}(\text{O}-t\text{-Bu})_3$ (vi) aniline

QUESTION FOUR [20 MARKS]

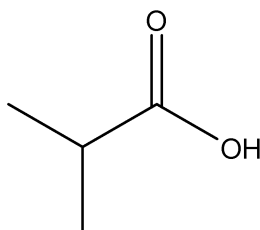
(a) Draw the structures of the following carboxylic acids **(3 marks)**

- (i) 4-aminopentanoic acid (ii) *trans*-2-methylcyclohexanecarboxylic acid
 (iii) 4-oxoheptanoic acid (iv) 3-chloroheptanedioic acid
 (v) 3,4-dimethylhex-3-enoic acid (vi) 3-methylphthalic acid

(b) Arrange the following compounds in order of increasing acidity and explain the basis for the chosen order. **(3 marks)**

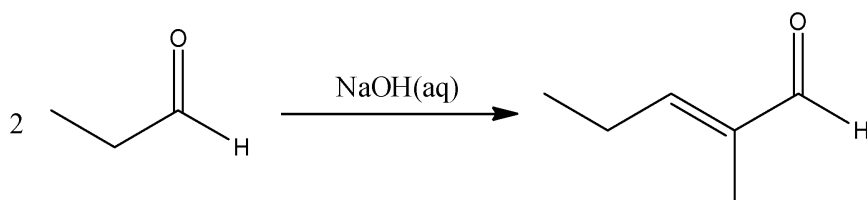


(c) Design a malonic ester synthesis for 2-methylpropanoic acid. **(4 marks)**

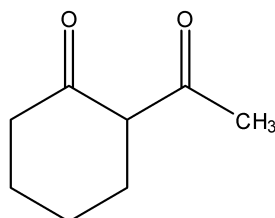


2-methylpropanoic acid

(d) Write the stepwise mechanism of the following reaction. **(4 marks)**



(e) Design a Stork enamine synthesis for the following compound starting with cyclohexanone and any other reagents of your choice. **(3 marks)**



(f) Write the stepwise reaction for the following reaction **(3 marks)**

