

## CHEM 436: ADVANCED STEREOCHEMISTRY AND REACTION MECHANISMS

STREAMS: B.ED (SC), B.SC
TIME: 2 HOURS
DAY/DATE: THURSDAY 7/12/2017
8.30 A.M - 10.30 A.M

## INSTRUCTIONS:

- Answer Question ONE and any other TWO Questions
- Do not write on the question paper


## QUESTION ONE: [30 MARKS]

(a) Name the following molecules according to R and S and IUPAC system of nomenclature.
(b) Draw the structures of the products of the following reactions (showing mechanisms in each case)
[8 Marks]
(c) Which sigmagrophic rearrangement do the following reactions represent?
(d) Draw a meso compound of the following molecules using Fisher's projection.
(ii) 1,2-dimethlcyclohexane
(iii) 1,3-diiodocyclopentane
(e) The enantiomeric excess of one compound in a mixture of a pair of enantiomer is $67.5 \%$. how much of each enantiomer is present?
(f) A pure compound has a rotation of $+13.2^{\circ}$. If a sample has a specific rotation of $2.64^{\circ}$ what is enantiomeric excess of this sample?
[2 Marks]

## QUESTION TWO [20 MARKS]

(a) (i) Draw ordinary structures and Newman projection of chair and boat conformers of cyclohexane and clearly show flagpole hydrogens where applicable.
(ii) Discuss stability of chair and boat conformers.
(b) Discuss conformations of disubstituted cyclohexanes (with relevant structures)
[10 Marks]
QUESTION THREE [20 MARKS]
(a) With clear illustrations of a 4 n system, describe how molecular orbitals are formed and show ground state and excited state configuration.
(b) Trans, cis, trans-2, 4, 6-octatriene (a $4 \mathrm{n}+2$ system) undergoes electrocyclic reaction. Show clear mechanisms how this works under thermal and photochemical conditions. [8 Marks]
(c) Briefly describe the two modes of orbital overlap during formation of two sigma bonds.
[4 Marks]

## QUESTION FOUR [20 MARKS]

(a) Exposure to uv light causes skin cancer. Explain how this happens and why the problem is not too much widespread.
(b) With examples, discuss classification of isomers.

