CHEM 436

CHUKA



UNIVERSITY

UNIVERSITY EXAMINATIONS

FOURTH YEAR EXAMINATION FOR THE AWARD OF BACHELOR OF EDUCATION (SCIENCE) AND BACHELOR OF SCIENCE

CHEM 436: ADVANCED STEREOCHEMISTRY AND REACTION MECHANISMS

STREAMS: B.ED (SC), B.SC

TIME: 2 HOURS

DAY/DATE: THURSDAY 7/12/2017

8.30 A.M - 10.30 A.M

INSTRUCTIONS:

- Answer Question ONE and any other TWO Questions
- Do not write on the question paper

QUESTION ONE: [30 MARKS]

(a) Name the following molecules according to R and S and IUPAC system of nomenclature.

[10 Marks]

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(b) Draw the structures of the products of the following reactions (showing mechanisms in each case) [8 Marks]

(c) Which sigmagrophic rearrangement do the following reactions represent? [5 Marks]

(d) Draw a meso compound of the following molecules using Fisher's projection. [3 Marks]

- (i) 3, 4-dichlorohexane
- (ii) 1,2-dimethlcyclohexane
- (iii) 1,3-diiodocyclopentane

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(e)	The enantiomeric excess of one compound in a mixture of a pair of enan	ntiomer is 67.5%.		
	how much of each enantiomer is present?	[2 Marks]		
(f)	A pure compound has a rotation of $+13.2^{\circ}$. If a sample has a specific rotation of $+13.2^{\circ}$.	tion of 2.64° what		
	is enantiomeric excess of this sample?	[2 Marks]		
QUESTION TWO [20 MARKS]				
(a)	(i) Draw ordinary structures and Newman projection of chair and bo	at conformers of		
	cyclohexane and clearly show flagpole hydrogens where applicable.	[5 Marks]		
	(ii) Discuss stability of chair and boat conformers.	[5 Marks]		
<i>a</i> \				

(b) Discuss conformations of disubstituted cyclohexanes (with relevant structures) [10 Marks]

QUESTION THREE [20 MARKS]

- (a) With clear illustrations of a 4n system, describe how molecular orbitals are formed and show ground state and excited state configuration. [8 Marks]
- (b) Trans, cis, trans-2, 4, 6-octatriene (a 4n+2 system) undergoes electrocyclic reaction. Show clear mechanisms how this works under thermal and photochemical conditions. [8 Marks]
- (c) Briefly describe the two modes of orbital overlap during formation of two sigma bonds.

[4 Marks]

QUESTION FOUR [20 MARKS]

(a)	Exposure to uv light causes skin cancer.	Explain how this happens	s and why the	problem is
	not too much widespread.		l	[10 Marks]
(b)	With examples, discuss classification of i	somers.		[10 Marks]

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