**CHUKA** 



**UNIVERSITY** 

# **UNIVERSITY EXAMINATIONS**

### EXAMINATION FOR THE AWARD OFBACHELOR OF SCIENCE AND BACHELOR OF EDUCATION SCIENCE

### PHYS 415: QUANTUM MECHANICS II

### STREAMS: BSC, BED (SCIENCE)

#### TIME: 2 HOURS

## DAY/DATE: THURSDAY 13/12/2018

11.30 A.M -1.30 P.M

#### **INSTRUCTIONS:**

- Answer question ONE and any other TWO questions.
- Clearly show your working.

### The following constants may be useful.

Gravitational constant G =  $6.672 \times 10^{-11} \text{N m}^2 \text{ kg}^{-2}$ 

Permittivity of free space  $\varepsilon_0 = 8.85 \ x \ 10^{-12}$ 

Boltzman constant  $k_B = 1.38 x 10^{-23} J/K$ =8.63 x 10-5 eV/K)

Planks constant  $h = 6.63 \times 10^{-34}$  J.S

Mass of an electron =  $9.1 \times 10^{-31} \text{ kg}$ 

Mass of a proton =  $1.67 \times 10^{-27} \text{ kg}$ 

Electronic charge =  $1.6021 \times 10^{-19}$ C

Reduced Plancks constant  $\hbar = 1.05 x 10^{-34} J. S.$ 

$$\int_{-\infty}^{\infty} e^{-ax^2} dx = \sqrt{\frac{\pi}{a}}$$
$$\int \sin^2 \theta d\theta = \frac{\theta}{2} - \frac{\sin 2\theta}{4}$$

#### **OUESTION ONE**

- (a) Giving one example in each case, state how particles and observables are represented in Quantum Mechanics. (2 marks)
- (b) A vector A, in Cartesian space takes the form  $\vec{A} = iA_x + jA_y + kA_z$  write this vector in Ket notation. (2 marks)
- (c) (i) Determine whether the matrix below is suitable in Quantum Mechanics. (3 marks)

$$\begin{bmatrix} 0 & \frac{-i\hbar}{2} \\ \frac{i\hbar}{2} & 0 \end{bmatrix}$$

(2 Marks)

Explain your answer in b (i) above. (2 marks) (ii) (3 marks)

(d) Briefly explain what is involved in the perturbation theory.

- (e) Describe the Stark effect.
- (f) Calculate the probability of finding a hydrogen atom electron within the radius of  $a_0$  given that its ground state wave function is;

$$\varphi_o(r) = \frac{1}{\sqrt{\pi a_0^3}} e^{\frac{-r}{a_0}}$$
(6 marks)

- (g) Calculate the relative energy shift between the unperturbed Is state and the corrected Is state due to the gravitational potential between the proton and the electron in the hydrogen atom. (6 marks)
- (h) Describe the Zeeman effect. (2 marks) (i) State Pauli's exclusion principle (2 mark)

#### **QUESTION TWO**

- Distinguish between degenerate and non-degenerate states. Give one example in (a) each case. (4 marks)
- Show that the amplitude  $a_m$  of the 1<sup>st</sup> order perturbation for the non-degenerate case can be given as;  $a_m = \frac{\langle \Psi m | H' | \omega_0 \rangle}{E_0 E_m}$  (8 marks) (b)
- The perturbation for a harmonic oscillator is  $H' = ae^{-\beta x^2}$ . Determine the energy (c) correction for the ground state energy. Take  $|\Psi_0\rangle = (\alpha/\sqrt{\pi})^{\frac{1}{2}}e^{-\alpha^2\frac{x^2}{2}}$ . (8 marks)

### **OUESTION THREE**

- (a) Derive the expression for the first energy correction resulting from first order perturbation. (8) marks)
- (b) Calculate the corrected second eigenvalue to first order perturbation in the finite well problem. Take  $\Psi_n = \sqrt{\frac{2}{L}} Sin \frac{2\pi x}{L}$ (12 marks)

#### **QUESTION FOUR**

 (a) (i) Distinguish between the approach taken by the Perturbation theory, variation method and WKB methods in approximating solutions to the Schrodinger equation. (4marks)

(ii) Using variation theory show that the eigenvalue with an unknown wave function  $\overline{\psi}$  gives the upper bound to the energy of the known wave function  $\psi$ . (5 marks)

(b) The Hamiltonian of a quantum oscillator can be given as  $H = -\frac{\hbar^2}{2m}\nabla + \frac{1}{2}m\omega^2 x^2$ , in one dimension. Calculate the lowest energy. (11 marks)

### **QUESTION FIVE**

(a) Work out the following commutation relations:-

	(i) $[L_x, L_y]$ (2 marks)	
	(ii) $[L_x, L_z]$	(2 marks)
	$(iii)[L_y, L_z]$	(2 marks)
(b) The states in a hydrogen atom has the principal quantum number n=4. Calculate t		
	(i) Corresponding angular momentum quantum number	(2 marks)
	(ii) Angular momentum L (2 marks)	
	(iii)Component of $\mathbf{L}$ in the z direction	(2 marks)
	(iv)Angle between L and z direction	(2 marks)
(c)	The Helium atom has more than one electron. Set up its Hamiltonian, taking this fact into	
	account and clearly explaining each term.	(6 marks)

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