

**CHUKA**



**UNIVERSITY**

**UNIVERSITY EXAMINATIONS**

**EXAMINATION FOR THE AWARD OF DEGREE OF BACHELOR OF SCIENCE**

**NURS 113: MEDICAL BIOCHEMISTRY I**

**STREAMS:**

**TIME: 2 HOURS**

**DAY/DATE: TUESDAY 04/12/2018**

**8.30 A.M. – 10.30 A.M.**

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**INSTRUCTIONS:**

- Answer question 1 and any other TWO questions.

**QUESTION ONE (30 MARKS)**

- (i) Which of the following is not a colligative property of water
- (a) Vapour pressure
  - (b) Boiling point
  - (c) Osmotic pressure
  - (d) Osmosis
- (ii) What is the concentration of  $OH^-$  in a solution with  $H^+$  concentration of  $1.3 \times 10^{-4} M$
- (a)  $7.7 \times 10^{-10} M$
  - (b)  $7.8 \times 10^{-10} M$
  - (c)  $7.7 \times 10^{-10} M$
  - (d)  $6.8 \times 10^{-11} M$
- (iii) The following are ways of regulation of pyruvate dehydrogenase which one is not?
- (a) Product inhibition by acetyl CoA
  - (b) Elevated levels of NADH
  - (c) Increased NADH/NAD<sup>+</sup> ratio
  - (d) Increase in insulin concentration

- (iv) Which of the following is not function of the krebs cycle
- (a) Energy generation
  - (b) Provide  $CO_2$  for gluconeogenesis
  - (c) Fatty acid synthesis
  - (d) Glycogen digestion
- (v) Which of the following is not a polysaccharide
- (a) Starch
  - (b) Glycogen
  - (c) Sucrose
  - (d) Cellulose
- (vi) Indicate which of the following is not a function of a lipid
- (a) Efficient energy sources
  - (b) Source of heme
  - (c) Serve as thermal insulators
  - (d) Serve as precursors of hormones
- (vii) Which is not a function of cholesterol
- (a) Synthesis of bile salts
  - (b) Synthesis of steroid hormones
  - (c) Synthesis of vitamin D
  - (d) Synthesis of glycogen
- (viii) Which of the following is not an essential amino acid
- (a) Arginine
  - (b) Alanine
  - (c) Leucine
  - (d) Tryptophan
- (ix) Indicate which of the following is not a conjugated protein
- (a) Nucleoprotein
  - (b) Lipoprotein
  - (c) Phosphoprotein
  - (d) Globulin
- (x) Which of the following is not a denaturing agent of proteins
- (a) Temperature
  - (b) Pressure
  - (c) Mechanical sheer force
  - (d) Glucose levels

- (xi) The following is a metal ion required for enzyme activity. Which one is not
- (a)  $Ca^{2+}$
  - (b)  $K^{+}$
  - (c)  $Mg^{2+}$
  - (d)  $Pb^{2+}$
- (xii) Which of the following is not a classification of enzymes
- (a) Transterases
  - (b) Oxidoreductases
  - (c) Glyceradehyde-3-phosphate
  - (d) Hydrolases
- (xiii) Which of the following is not a factor that affects enzyme action.
- (a) Temperature
  - (b) pH
  - (c) Substrate/enzyme concentration
  - (d) Van der Waals forces
- (xiv) The following are methods of regulating enzyme activity which one is not?
- (a) Irreversible covalent modification
  - (b) Reversible covalent modification
  - (c) Allosteric modulation
  - (d) Alkaline phosphate
- (xv) Indicate which of the following enzymes is not used in clinical diagnosis
- (a) Lipase
  - (b)  $\alpha$  -amylase
  - (c) Trypsin
  - (d) Chryomotrypsin elastase
- (xvi) Indicate which is not a hexose
- (a) Glucose
  - (b) Ribose
  - (c) Galactose
  - (d) Mannose
- (xvii) Which of the following is not a physical property of a monosaccharide
- (a) Colourless
  - (b) Crystalline
  - (c) Readily soluble in  $H_2O$
  - (d) Coloured

- (xvii) Solutions of equal osmolality are
- (a) Hypertonic
  - (b) Isotonic
  - (c) Hypotonic
  - (d) Osmosis
- (xix) Hydrophilic substances are
- (a) Water loving
  - (b) Water hating
  - (c) Insoluble in water
  - (d) Water scarce
- (xx) Which of the following does not determine the intermediate to which pyruvate is converted
- (a) Types of cells
  - (b) Aerobic prokaryotes
  - (c) Anaerobic prokaryotes
  - (d) White blood cells
- (xxi) Which of the following enzyme is not part of pyruvate dehydrogenase complex
- (a) Pyruvate dehydrogenase
  - (b) Dihydrolipoyl transacylase
  - (c) Dihydroxyacyl hydrogenase
  - (d) Succinyl CoA
- (xxii) Which of the following is not a classification of carbohydrates
- (a) Monosaccharides
  - (b) Oligosaccharides
  - (c) Phospholipid
  - (d) Polysaccharide
- (xxiii) All the following proteins are soluble in  $H_2O$ . Which one is not?
- (a) Albumins
  - (b) Collagens
  - (c) Globulins
  - (d) Histones
- (xxiv) The following are ways in which metal ions promote enzyme action which one is not?
- (a) Maintaining or producing the active structural conformation of the enzyme
  - (b) Promoting the formation of the enzyme substrate complex
  - (c) Acting as electron donors or acceptors
  - (d) Causing distortions in the substrate or the enzyme

(xxv) Which of the following is not a disaccharide?

- (a) Maltose
- (b) Lactose
- (c) Sucrose
- (d) Starch

(xxvi) The following is not an example of a non-standard amino acid

- (a) 4-hydroxyproline
- (b) 5-hydroxylysine
- (c) Alanine
- (d) 6-N-methyl cysteine

(xxvii) Which of the following is not a conjugated protein

- (a) Nucleoprotein
- (b) Lipoprotein
- (c) Phosphoprotein
- (d) Globulin

(xxviii) The following enzymes have been matched to the specific bond they catalyze. Which one is not?

- (a) Esterases-Esterbond
- (b) Peptidases -peptide bonds
- (c) Glycosidases-Glycosidic bonds
- (d) Trypsinogen-psinogen bonds

(xxix) Which of the following enzyme is not involved in glycolysis

- (a) Hexokinase
- (b) Phosphohexose isomerase
- (c) Phosphofuctokinase
- (d) Insulin

(xxx) Which of the following is not an electron carrier

- (a)  $NAD^+$
- (b)  $NADP^+$
- (c) FAD
- (d) Pyruvate

**QUESTION 2 (20 MARKS)**

- (a) Explain the relevance of measurement of pH in medical diagnosis (2 marks)
- (b) Define the following terms
- (i) Buffers
  - (ii) Osmosis
  - (iii) Free energy
- (c) Energy released upon hydrolysis of high energy phosphate bond may result in the following. (3 marks)
- (d) Draw and label the mitochondria (3 marks)
- (e) Differentiate between
- (i) Aldoses and ketoses
  - (ii) Homopolysaccharides and heteropolysaccharides
  - (iii) Simple lipids and complex lipids (6 marks)
- (f) Give three functions of phospholipids (3 marks)

**QUESTION 3 (20 MARKS)**

- (a) Explain the mechanisms in animals and plants that prevent osmotic lysis. (3 marks)
- (b) Calculate the pKa of lactic acid given that when the concentration of lactic acid is 0.010M and the concentration of lactate is 0.087M, the pH is 4.80. (4 marks)
- (c) Define the following
- (i) Oxidation
  - (ii) Reduction
  - (iii) Exergonic
  - (iv) Endergonic (4 marks)
- (d) Give three functions of glycosaminoglycans (3 marks)
- (e) Differentiate between
- (i) Ketogenic and glucogenic amino acids
  - (ii) Essential and non-essential amino acids (4 marks)
- (f) Draw the general formula of a naturally occurring amino acid. (2 marks)

**QUESTION 4 (20 MARKS)**

- (a) Briefly discuss the four levels of organization of proteins. (4 marks)
  - (b) Give two properties of a denatured proteins. (2 marks)
  - (c) Define the following terms
    - (i) Apoenzyme
    - (ii) Cofactor
    - (iii) Coenzyme
    - (iv) Prosthetic group (4 marks)
  - (d) Discuss three properties of enzymes (6 marks)
  - (e) Discuss briefly the Michaels and Menten hypothesis of enzyme action. (2 marks)
  - (f) Differentiate between irreversible and reversible enzyme inhibition. (2 marks)
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