CHEM 344

CHUKA



UNIVERSITY

UNIVERSITY EXAMINATIONS

THIRD YEAR EXAMINATION FOR THE AWARD OF DEGREE OF BACHELOR OF SCIENCE

CHEM 344: ENVIRONMENTAL CHEMISTRY I

STREAMS:

TIME: 2 HOURS

DAY/DATE: TUESDAY 30/03/2021

2.30 P.M – 4.30 P.M

INSTRUCTIONS:

Answer question one and any other two questions

QUESTION ONE (30 MARKS)

re varies with	
es between	
[3 marks]	
hat are the	
[3 marks]	
b) (i) Briefly explain why temperature decreases with altitude in the troposphere, nut	
[3 marks]	
of	
[1 mark]	
(iii) List the names, the chemical formulae and sources of any three major gaseous	
[4 marks]	
[3 marks]	
depleted in the	
[3 marks]	
[1mark]	

(iii) What are the consequences of the depletion of O_3 layer in the atmosphere? [2 marks]

(iv) The atmosphere plays an essential role as a protective shield. Briefly describe how this is achieved. [2 marks]

(d) (i) Distinguish between photodissociation and photoionization. [2 marks] (ii) Given that the bond energy is oxygen is 495KJ mol^{-1} . What is the maximum wave length of light in nanometers (nm) that has enough energy per photon to dissociate the oxygen molecule?

[Planck's constant h= 6.626×10^{-34} J-s Avogadro's number N= 6.022×10^{23} /mol, speed of light, c = 3.0×10^8 m/s. [3 marks]

QUESTION TWO (20 MARKS)

(a) (i) With help of relevant equations, give a detailed explanation on how photochemical smog is formed. [3 marks]
(ii) What are the environmental conditions required to form photochemical smog? [1 mark]

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- (iii) What are harmful effects of photochemical smog and how can they be controlled? [3 marks]
- (b) (i) $N_{2,}O_{2}$ and O absorb photons in the upper atmosphere. Explain why these processes don't completely protect us from UV radiation.

[2 marks]

(ii) Write the chemical reactions that describe the cycling of NO and NO_2 to maintain their relative concentrations in the atmosphere assuming no hydrocarbons are present.

[2 marks]

(c) (i) Explain in details what greenhouse effect is and how it affects the global climate.

[4

marks]

(ii) With respect to absorption of radiant energy, what distinguishes a green house gas from non- green house gas?[2 marks]

(iii) Explain using molecular structure of CO_2 why it is a green house gas but Ar is not . Name any other two green house gases. [3 marks]

QUESTION THREE (20 MARKS)

(a) (i) What are CFCs chemicals? What properties of CFCs make them ideal for various commercial applications but also make them a long term problem in the stratosphere?

marks]

(ii) Using CF_2CL_2 as an example, show how CFCs reactions are involved in ozone depletion in the stratosphere. [2 marks]

[3

- (b) (i) What are hydrofluorocarbon? Why are these compounds potentially less harmful to the ozone layer than CFCs. [3 marks]
- (c) (i) Describe acid rain? [2 marks]
 (ii) Name and give sources of the pollutants responsible for causing acid rain. [3 marks]
 (iii) Explain how acid rain harmful to the environment? Mention any three practical ways that can mitigate the problem of acid rain. [4 marks]

QUESTION FOUR

- (a) Distinguish between the following pairs: [2 marks]
- (i) Pollutants and contaminant
- (ii) Primary pollutants and secondary pollutants
- (b) (i) What are the major primary air pollutants produced by human activity? What are their sources? [4 marks]
 (ii)Briefly discuss the impact of air pollution on human health and by giving examples where necessary, highlight on three major health impacts of air pollutants. [4 marks]
- (c) (i) Briefly explain the major causes of water pollution. [3 marks]

(ii)Name two common methods of secondary treatment of sewage. Briefly describe each of these two methods. [4 marks]

(d) Explain the meaning of green chemistry? How will it help in decreasing? Environmental pollution? [3 marks]
