CHEM 103

CHUKA



UNIVERSITY

UNIVERSITY EXAMINATIONS

EXAMINATION FOR THE AWARD OF DEGREE OF BACHELOR OF SCIENCE

CHEM 103: GENERAL ORGANIC CHEMISTRY

STREAMS: BSC TIME: 2 HOURS

DAY/DATE: TUESDAY 06/04/2021 8.30 A.M – 10.30 A.M.

INSTRUCTIONS: Answer QUESTION ONE and any OTHER TWO questions.

QUESTION ONE (30 MARKS)

a) i) Explain hybridization briefly

(2 marks)

- ii) Using orbital diagrams discuss the hybridization of methane (4 marks)
- b) Explain why the C-H bond lengths of Ethyne is 1.06 Å, ethene is 1.09 Å and ethane 1.10 Å (2 marks)
- c) Define

i) Catenation (1 mark)

ii) Functional group (1 mark)

d) Using an equation give two methods for preparation of alkanes (4 marks)

e) Draw the structures of the following molecules (5 marks)

- i) 3-methylhexane
- ii) 2,3-dimethylheptane
- iii) 3-ethyl-2,4-dimethyloctane
- iv) 2,3-dimethyloctane

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f) Give the IUPAC names of the following structures

(5 marks)

- i) CH₃CH=CH₂
- ii) CH₃CH(CH₃)CH₂=CH₂
- iii) CH₃CH₂CH=C(CH₂)CH(Cl)CH₃



- iv)
- v) CH₃CH₂CH(CH₃)CH₂OH
- g) Complete the following reactions giving the major product(s) A, B, C and D

(4

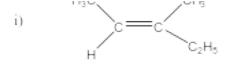
marks)

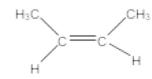
ii) CH3CHBrCHBrCH₃ Zn ► B

iii) CH₃C = CCH₃ H₂/Lindlar catalyst

- iv) CH₂−CH₂ 1.BH₂/H₂O₂ D
- h) Assign the E/Z nomenclature of the following alkenes

(2 marks)





QUESTION TWO (20 MARKS)

a) Give the IUPAC name of the following alkanes

(5 marks)

- i) CH₃CH₂CH(CH₃)CH(CH₃)CH₃
- - iv) CH₃(CH₂)₅CH₃
 - CH3 CH₃
 CH₃CH₂CHCHCHCH₃
 CH₂CH₃
 CH₃CH₃CH₃CH₃CH₃CH₃CH₃
- b) Draw the mechanism of the following reaction

(5 marks)

c) Using an equation give two methods for synthesis of alkene

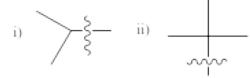
(4 marks)

d) Draw the following hydrocarbons

(4 marks)

- Cyclopropane i)
- ii) Cyclobutane
- 1,3-dimethylcyclohexane iii)
- 3-cyclobutylpentane iv)
- e) Name the following substituents

(2 marks)



QUESTION THREE (20 MARKS)

a) State three physical properties of alkanes

(3 marks)

b) Name the major product(s) of the following reactions

(5 marks)

i)
$$CH_3C$$
- $CHCH_3$ $1. O_3$ $A + B$ CH_3 CH_3

iii)
$$CH_3CH_2C$$
 CCH_2CH_3 Na/NH_3 E

c) i) Define isomers

(1 mark)

ii) Write the isomers of with the molecular C₆H₁₄

(5 marks)

d) (i) Define Markonikov's rule

(1 mark)

ii) Give the products of the following reactions

(3 marks)

i)
$$H_3C$$
 $+$ HCl A

ii)
$$H_3C$$
 H_2O_2 H_3C

- e) Name the following compounds according to IUPAC system of nomeclature (2 marks)
 - j) CH₃CH(OH)CH₃
 - ii) CH₃CH₂CH(CH₃)CH₂C(C₂H₅)CH₂CH₂OH

QUESTION FOUR (20 MARKS)

a) Name the following compounds according to IUPAC system of nomenclature (5 marks)

b) Write a method for laboratory differentiation between an aldehyde and a ketone

(2

marks)

c) Draw the structure for the following compounds

(5 marks)

- i) Methanal
- ii) 2-methylhexanal
- iii) Ethylamine
- iv) 2-methyl-1-propanamine
- v) Diethyl ether

d) Using a suitable equation give two methods of preparation of alkynes (4 marks)

e) Design a stepwise method for synthesis of pent-2-yne starting from ethyne and any other reagents (4 marks)
